

# Chapter 7 Ionic And Metallic Bonding

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1 chapter 7 "ionic and metallic bonding" click to add text 2 section 7.1 - ions objectives:-determine the number of valence electrons in an atom of a representative

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section 7.1 assessment. how can you determine the number of valence electrons in an atom of a representative element? atoms of which elements tend to gain electrons?

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section 7.1 - ions. valence electrons are the electrons in the highest occupied energy level. valence electrons are the only electrons involved in chemical bonding.

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the octet rule. in chapter 6, we learned that noble gases are unreactive in chemical reactions. in 1916, gilbert lewis used this fact to explain why atoms form ...

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formation of cations an atom is electrically neutral because it has equal numbers of protons and electrons; an ion forms when an atoms or group of atoms loses or ...

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chapter 7 7.1 ionic and metallic bonding valence electrons are the electrons in the highest ... chapter 7 ionic bonding keeping track of electrons the electrons ...

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section 7.1 assessment you have already answered these with your text questions. how can you determine the number of valence electrons in an atom of a representative ...

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chapter 7 ionic and metallic bonding65 14. what four properties make steel an important alloy? a. b. c. d. 15. what are the component elements for the following ...

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the octet rule. in chapter 6, we learned that noble gases are unreactive in chemical reactions. in 1916, gilbert lewis used this fact to explain why atoms form ...

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chapter 7 ionic and metallic bonding 167 chapter quiz classify each of these statements as always true, at; sometimes true, st; or never true, nt.

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